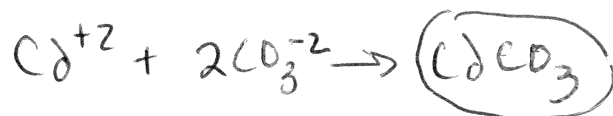
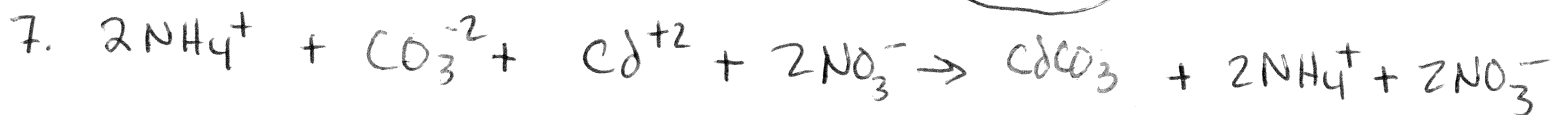
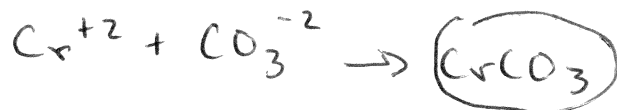
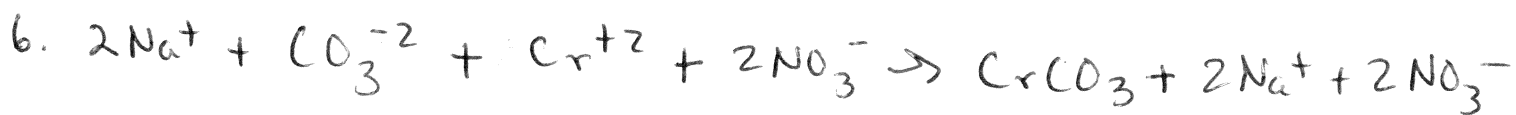
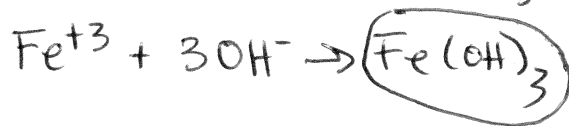
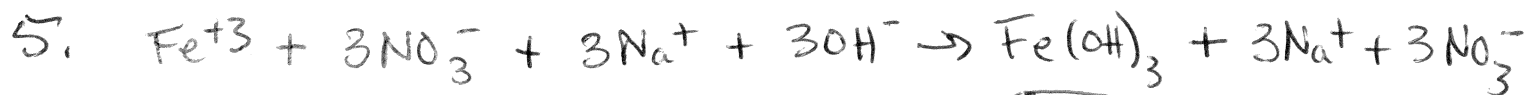
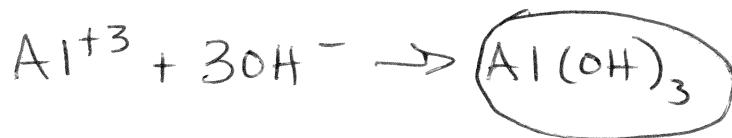
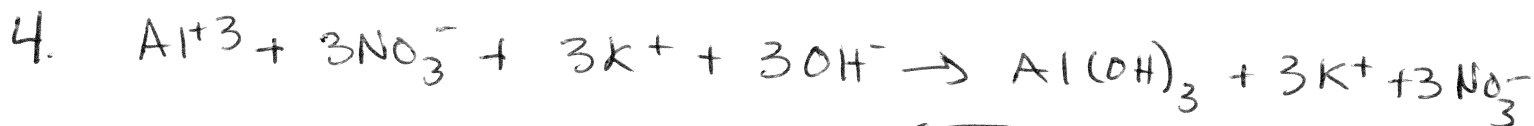
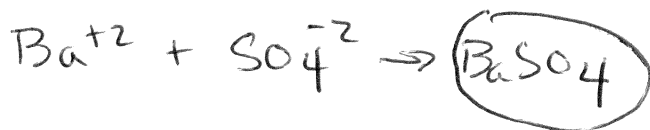
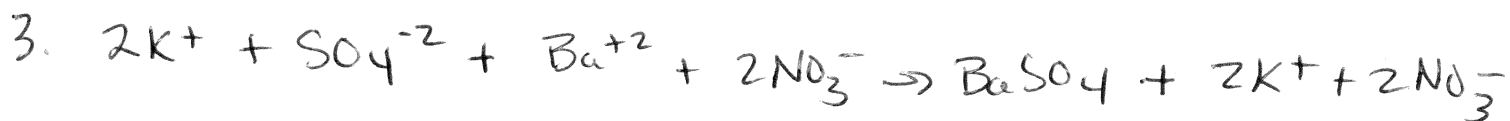
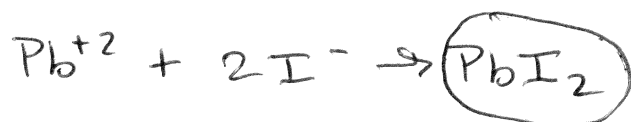
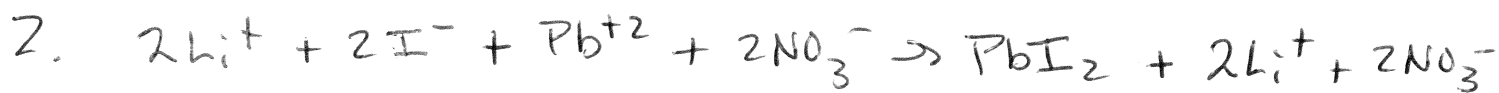
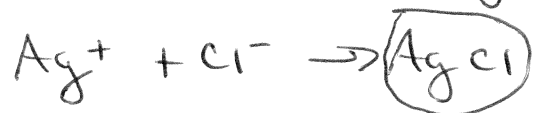
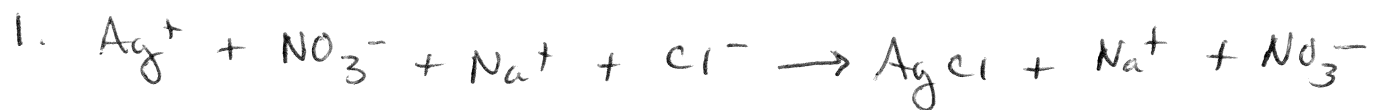
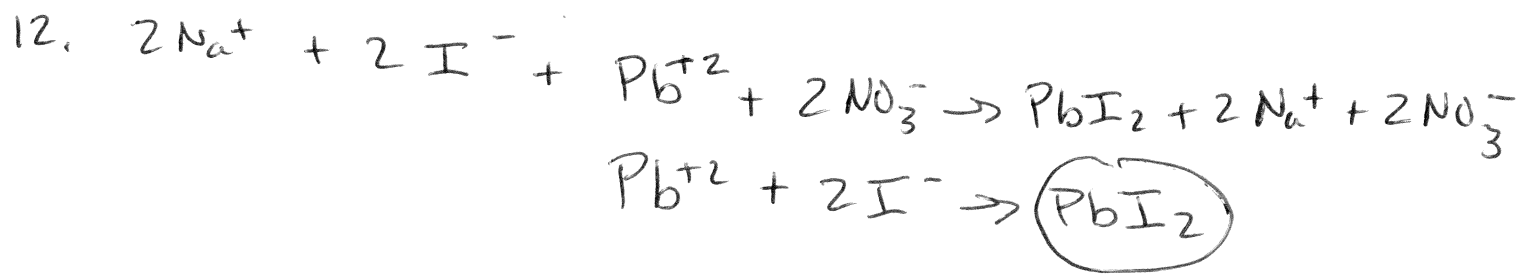
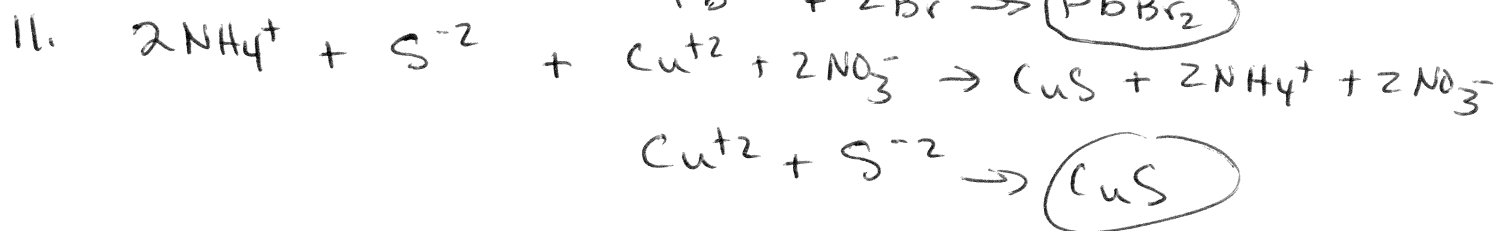
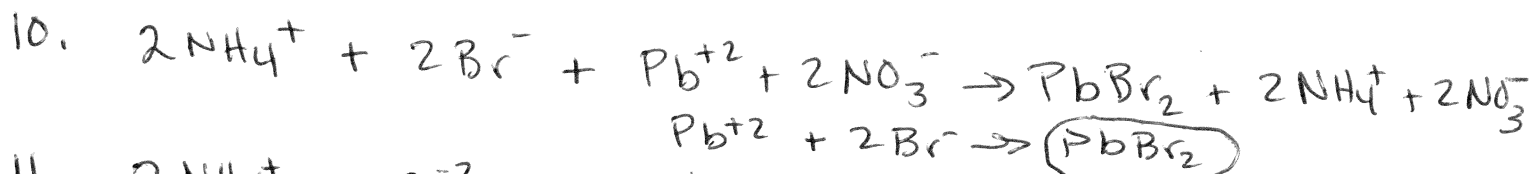
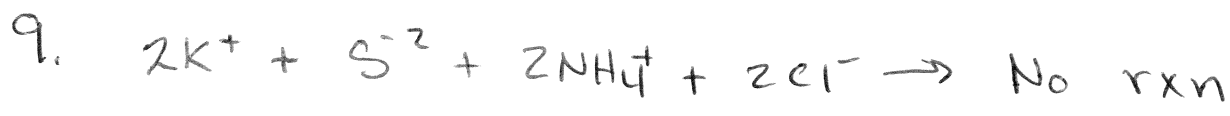
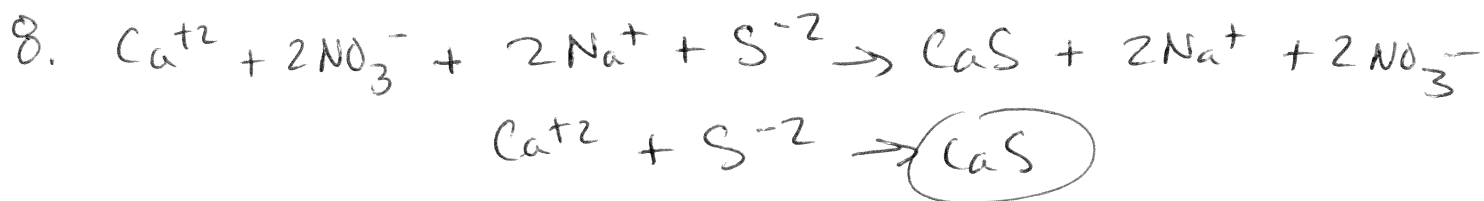


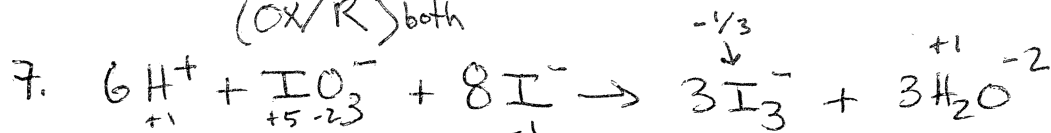
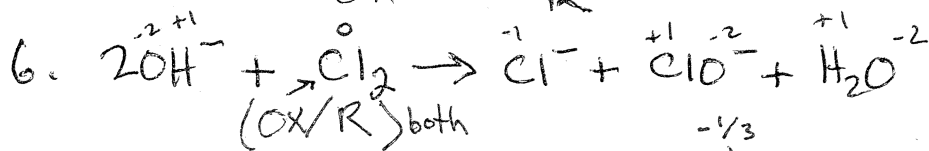
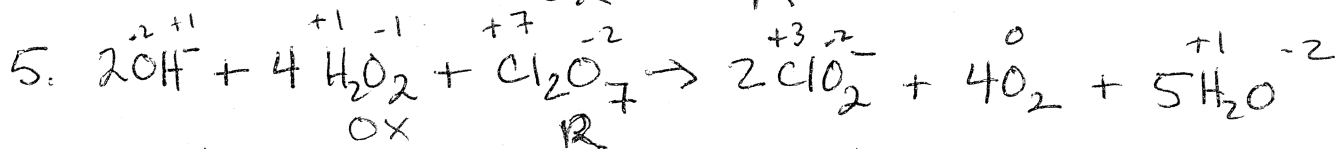
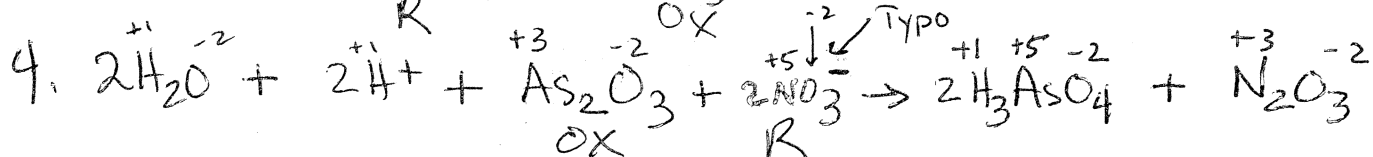
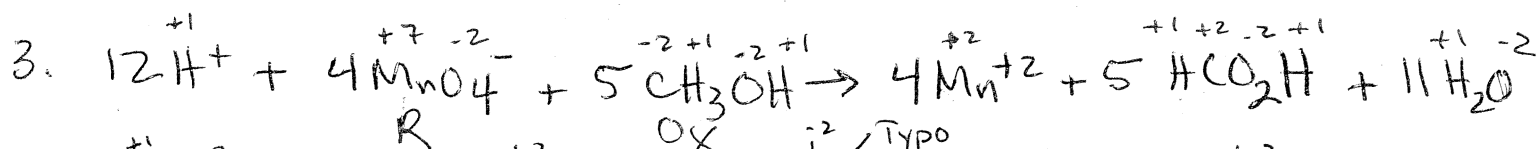
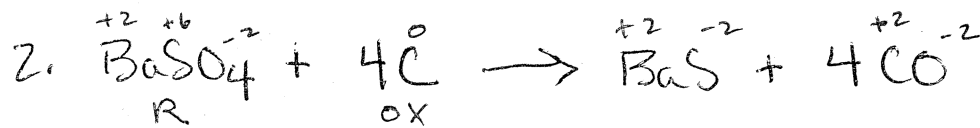
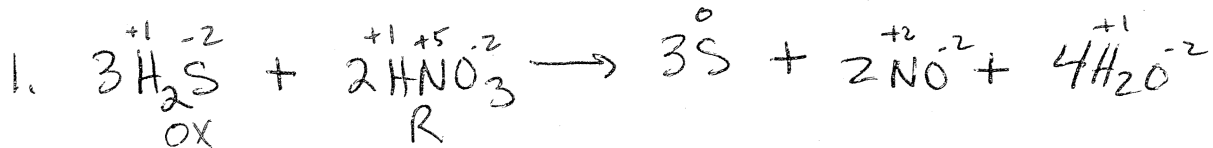
chem 0 Review Precipitation Rxns

Key



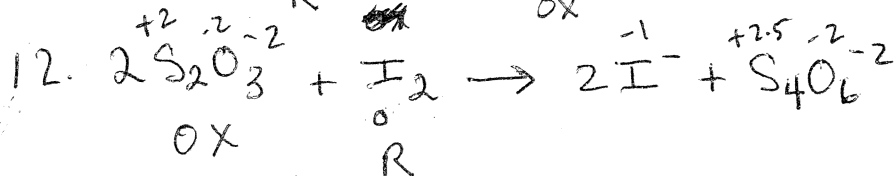
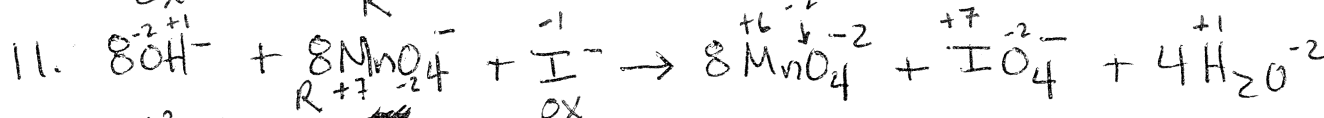
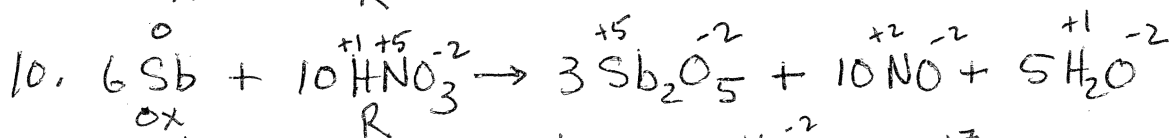
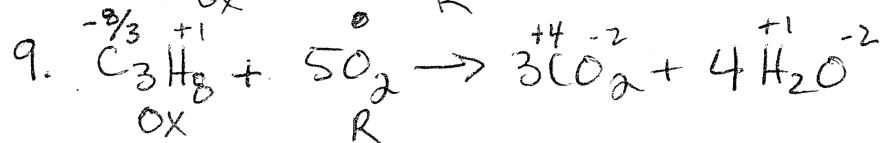
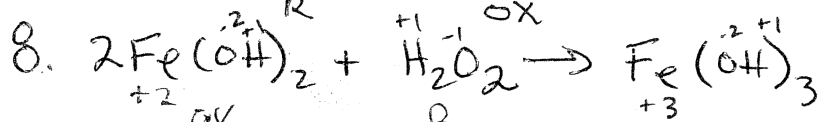


Determine if reaction is an oxidation-reduction reaction. Show oxidation numbers. Do work in notebook. Identify reactant oxidized and reduced.



$$3\text{I} = -1$$

$$\text{I} = -1/3$$



$$\text{C} - 4 = 0$$

$$\text{C} = +4$$